Prospecting for Lunar 3He with a Radio-Frequency Atomic Magnetometer

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I. K. Kominis^{1,2,3,a)} and C. Kosmas⁴

AFFILIATIONS

- Department of Physics, University of Crete, Heraklion 70013, Greece
- ²School of Automation and Electrical Engineering, Zhejiang University of Science and Technology, Hangzhou 310023, China
- ³Quantum Biometronics P.C., Heraklion 71409, Greece
- Lunar Cargo P.C., Athens 16343, Greece

ABSTRACT

Mining ³He from lunar regolith has attracted significant interest in recent years due to the scarcity of ³He on Earth and its diverse applications, from cryogenics and medical imaging, to nuclear physics and future nuclear fusion. Given the stringent technical and economic challenges of mining lunar 3 He, precise prospecting is essential. Here, we propose a prospecting methodology based on a radio-frequency atomic magnetometer, which can detect the dipolar magnetic field of thermally polarized 3 He spins. With a 200 g regolith sample and an rf magnetometer with sensitivity $1 \, \mathrm{fT/\sqrt{Hz}}$, we can detect 3 He with abundance 5 ppb within a measurement time of just 5 min. The associated apparatus is lightweight and significantly more cost-effective than alternative measurement techniques. The proposed prospecting method is readily deployable and could substantially improve the technical and economic feasibility of mining lunar 3 He.

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I. INTRODUCTION

The Moon is not only a stepping stone for space exploration, but also the home of valuable extraterrestrial resources.² Mining lunar ³He has attracted considerable interest, on the one hand because of its limited availability on Earth, and on the other because of its potential use in nuclear fusion.³ Unlike conventional D-T fusion, D-3He reactions produce minimal neutron radiation, enhancing power production efficiency while minimizing longterm radioactive waste. With its nuclear fuel potential aside, He is already used for low temperature physics and cryogenics,5 increasingly applied to emerging quantum technologies, 10,11 for magnetic resonance imaging, 12 and nuclear physics. 13,1

Lunar regolith blankets moon's surface and contains measurable quantities of ³He, at the level of 1-30 ppb, ¹⁵⁻¹⁷ implanted by the solar wind, ^{18,19} and measured by analyzing samples returned by the Apollo²⁰ and later missions.^{21,22} Over time, exposure to solar wind has led to the gradual accumulation of ³He, particularly in titanium-rich minerals, such as ilmenite,²³ with an estimated quantity around 10⁶ tons.³

Extraction of ³He proceeds by heating regolith to temperatures around 1000 K^{24,25} and separating, e.g., cryogenically, the released gases.²⁶ The low abundance of ³He requires processing of large regolith mass; hence, the extraction process would have to be performed on moon's surface. Given the extraction's non-trivial technical demands,²⁷ it is crucial to precisely prospect for ³He and identify areas with the highest abundance.

Here, we propose a direct measurement to detect regolithimplanted ³He by use of an atomic magnetometer, in particular, a radio-frequency magnetometer. The proposed method is able to detect lunar ³He from a 200 g regolith sample at the level of 5 ppb within a measurement time of 5 min, while consuming minimal power, and being light-weight. Thus, the proposed detection technique is readily deployable in the lunar environment. The cost of the relevant apparatus is negligible compared to mission costs and significantly less than alternative prospecting methodologies. Hence, from the economic perspective, the proposed technique enables swift prospecting campaigns with a compact device, potentially saving on mission duration, complexity, and cost.

a)Author to whom correspondence should be addressed: ikominis@uoc.gr

For completeness, we note that several authors^{28–31} have questioned the combined technical/economic viability of proposals for mining lunar ³He, further claiming that other earth-based sources could come online, such as ³He-breeding reactors. While such arguments are indeed sound, we note that there are intricate and many times surprising links between economy and technology. For example, the outlook of economic viability could change abruptly if the same infrastructure could be used for mining additional resources. In any case, we here opt to remain agnostic regarding the business case for mining lunar ³He and merely delve into the technical exercise of prospecting for this gas.

The structure of the paper is as follows. In Sec. II, we briefly discuss existing methodologies for detecting lunar ³He. In Sec. III, we present the possibility of using a radio-frequency (rf) atomic magnetometer. We conclude in Sec. IV.

II. EXISTING TECHNIQUES FOR QUANTIFYING LUNAR ³He

Indirect measurements are primarily based on remote sensing^{32–37} from lunar orbiters, most of which probe for titanium, which shows strong correlation with ³He abundance. While remote sensing methods provide valuable first-order estimates for identifying promising regions containing ³He, they are inherently limited in precision and reliability as they rely on correlations rather than direct measurements. Factors, such as regolith depth, surface exposure history, and the efficiency of solar wind implantation, introduce significant variability difficult to resolve. Thus, direct in situ measurements remain critical for any mission aiming to mine ³He at economically viable scales.

So far, such measurements rely mostly on mass spectrometers, $^{38-45}$ with several variants, such as quadrupole mass spectrometry, resonance ionization mass spectrometry, or time-of-flight mass spectrometry. Such techniques ionize atoms or molecules with different ionization schemes and then use electromagnetic fields to separate the resulting ions based on their mass-to-charge ratios. While they require very small sample mass and provide for highly sensitive ³He detection, at the level of 1 ppb, the necessary equipment is rather bulky, massive, and costly. For example, the mass spectrometer reported in Ref. 41 has weight over 10 kg and volume over 40 l. A miniaturized time-of-flight mass spectrometer for lunar water detection has a volume of 71.46 For comparison (see Sec. IV), the device proposed herein has an estimated volume of less than 1 l. In any case, we here wish to propose an alternative prospecting methodology. The particular methodology, or even a hybrid arrangement of several methodologies that shall be optimal, will be decided when designing a space mission taking into account numerous engineering constraints beyond mass or volume of the prospecting device (see Sec. IV).

III. MEASUREMENT WITH AN ATOMIC MAGNETOMETER

Atomic magnetometers^{47–55} detect magnetic fields by optically probing a spin-polarized alkali-metal vapor. The quantum state of the atoms in the vapor is influenced by the optical pumping and probing light, atomic collisions, internal atomic hyperfine interactions, and last but not least, by the magnetic field to be measured.

Spin-exchange-relaxation-free-magnetometers 56,57 have demonstrated sub-fT magnetic sensitivity at a zero background magnetic field. On the other hand, rf magnetometers^{58–66} work at a specific non-zero bias magnetic field and detect a weak ac magnetic field at the respective Larmor frequency. Such magnetometers utilize the fact that large spin polarization also suppresses spin-exchange relaxation, and additionally, working at high frequencies alleviates technical noise, such as magnetic noise produced by the material used to magnetically shield the former devices.

The general idea of the proposed measurement is the following. Lunar ³He shall be extracted from regolith, spin-polarized, and captured in a small measurement cell. A free-induction decay will then be induced.⁶⁷ The precessing spins of the magnetized ³He vapor will produce a dipolar magnetic field oscillating at the precession frequency. The rf magnetometer will detect this ac field, as in atomic-magnetometer-detected nuclear magnetic resonance.^{68–71} The envisioned measurement setup is shown in Fig. 1. Parenthetically, performing a similar measurement in the solid regolith sample would lead to rather prohibitive spin relaxation times for ³He, and this is why we need to have ³He be released from regolith and measured in the gas phase.

A. Number of ³He atoms captured in the measurement cell

In more detail, lunar ³He shall be extracted by heating a small regolith sample at $T_h=1000\,\mathrm{K}.^{72}$ Some of the released gases, such $_{\mathrm{S}}$ as H₂O, SO₂, or H₂S, will be captured by a cold trap at a temperature, e.g., $T_p = 100 \text{ K}$. The rest of the gases not liquifying will diffuse toward the measurement volume. Such gases include ⁴He, H₂, CO, CO₂, and N₂. For this prospecting measurement, there is no need to separate them from ³He, as they are nonmagnetic and their effect on ³He spin-relaxation time is negligible at the low pressure of this measurement. pressure of this measurement.

Now, we consider the measurement cell of volume $V_{\rm He}=1\,{
m cm^3}$, also being at temperature T_p . Within the temperature gradient defined by the heating temperature T_h and the measurement cell temperature T_p , the capture efficiency C of $^3\mathrm{He}$ atoms by the measurement cell should $\mathcal{C} = (V_{\mathrm{He}}/V_p)\sqrt{T_h/T_p}/(V_h/V_p + \sqrt{T_h/T_p})$, where V_h and V_p are the gas volumes at temperatures T_h and T_p , respectively. Noting that $V_p = V_{\rm gh} + V_{\rm He}$, where $V_{\rm gh}$ is the volume of the gas-handling system (also at temperature T_p) other than the measurement cell, and assuming $V_{\rm gh} = V_{\rm He}$ and $V_p = V_h$, it follows that $\mathcal{C} \approx 38\%$.

Given a regolith sample of mass $m_s = 200 \,\mathrm{g}$ and density $1.5 \,\mathrm{g/cm^3}$, and denoting by \tilde{m} (with unit ppb) the ³He mass abundance, the resulting ³He atom number in the measurement cell will

$$N_{\text{He}} \approx 2 \times 10^{16} \left[\frac{\mathcal{C}}{38\%} \right] \left[\frac{\tilde{m}}{1 \text{ ppb}} \right] \left[\frac{m_{\text{s}}}{200 \text{ g}} \right].$$
 (1)

B. Thermal spin polarization of ³He

Before entering the measurement cell, ³He shall be prepolarized inside a polarizing magnetic field $B_p = 1 \,\mathrm{T}$ and at

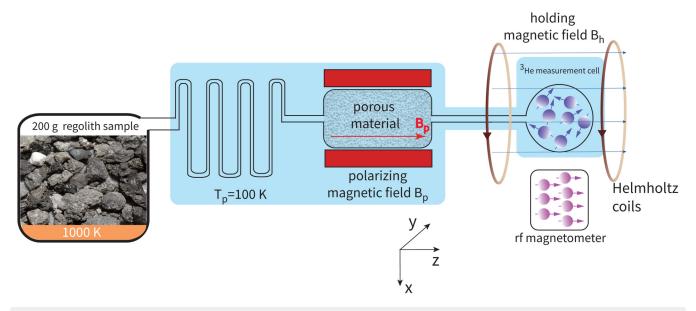


FIG. 1. Schematic (not drawn to scale) apparatus for measuring the abundance of 3 He in regolith. A regolith sample of mass $m_{\rm s}=200\,{\rm g}$ is heated, and the extracted gases flow through a cold trap at $T_p = 100 \,\text{K}$. Then, follows a polarizing magnetic field $B_p = 1 \,\text{T}$, where ³He is thermally spin-polarized by flowing through a porous material designed to match diffusion time with longitudinal spin-relaxation time. Finally, the gas enters the measurement cell inside a smaller and homogeneous holding magnetic field $B_h = 10$ G. A free-induction decay is induced, and the resulting oscillating magnetic dipolar field produced by 3 He spins is detected by the radio-frequency magnetometer.

temperature T_p . Such magnetic field can be readily established with small permanent magnets in a Halbach design. 74,75 In this step, ³He spins will be thermally polarized, attaining polarization $P_{\rm He} \approx \mu B_p/k_B T_p$, where $\mu = 1.07 \times 10^{-26} \, {\rm J/T}$ is the nuclear magnetic moment of ³He and $k_B = 1.38 \times 10^{-23}$ J/K is Boltzmann's constant. It follows that

$$P_{\text{He}} \approx 8 \times 10^{-6} \frac{[B_p/1 \text{ T}]}{[T_p/100 \text{ K}]}.$$
 (2)

During this pre-polarization step, ³He will be diffusing through a porous material,⁷⁶ which serves to slow down the diffusion of ³He atoms and, thus, increase the time they spend in the polarizing magnetic field. The transit time through this material should be larger than the longitudinal relaxation time T_1 . This can be accommodated by materials such as aerogels.⁷⁷ For example, consider the sol-gel used to coat spin-polarized ³He glass cells.⁷⁸ In one example, for a 2 atm cell having 2.5 cm diameter, the longitudinal relaxation time was $T_1 \approx 300 \, \text{h}$. The ³He self-diffusion coefficient at this pressure and at room temperature is about⁷⁹ 1 cm²/s; thus, ³He atoms collide with the cell walls about 10⁶ times within the time T_1 . For a porous cylinder of length L=1 mm and diameter 1 mm, made of the same material and having pores of diameter $2a = 100 \,\mathrm{nm}$, ³He atoms will collide with the material about 10^8 times before exiting; thus, there is enough time for the ³He spin polarization to equilibrate at the value P_{He} .

Indeed, in the idealized case of straight cylindrical pores and diffuse wall scattering, the Knudsen diffusion coefficient is $D_K = \frac{2}{3} a \bar{\nu}$, where $\bar{\nu} = \sqrt{8k_B T_p / \pi m} \approx 10^3 \text{ m/s}$. $D_K \approx 3 \times 10^{-5} \,\mathrm{m}^2/\mathrm{s}$. The nominal transit time across the length $L \approx 3 \times 10^{-5} \,\mathrm{m}^2/\mathrm{s}$. is $t_0 \approx L^2/2D_K \approx 20$ ms, and the wall-hit frequency is $\aleph^{10}_{\rm N} \approx 10^{10} \, {\rm s}^{-1}$, so the number of collisions with the sol-gel is $v_{\rm wall} \, t_0 \approx 4 \times 10^8$, as stated in the previous paragraph.

In a realistic porous medium of porosity (void fraction) $p \le 1$ and tortuosity (squared ratio of actual path length to straight length) $\tau \ge 1$, the effective diffusion constant becomes $D_{\rm eff} = (p/\tau)D_K$, and the transit time and collision count change accordingly. For example, for p = 0.5 and $\tau = 3$, the time t_0 becomes 100 ms and the number of collisions 20×10^8 . In any case, the transit time is negligible compared to the magnetometric measurement time to be discussed later. Finally, the relaxation due to collisions with the glass wall containing the porous material is negligible when the porous material tightly fits the glass cylinder and in any case would reduce the relaxation time, which would further ensure the thermal equilibrium spin polarization of ³He.

The measurement cell resides in a holding magnetic field $B_h < B_p$, e.g., $B_h = 10$ G. This is because it is less straightforward to create a strong magnetic field for volumes large enough to accommodate the measurement cell. Thus, we opt to pre-polarize ³He in the "large" polarizing field B_p . Then, the measurement can proceed in a lower, albeit homogeneous holding magnetic field B_h . The rf magnetometer resides in a smaller magnetic field, chosen so that the Larmor resonance of the employed alkali-metal atom coincides with the precession frequency of the ³He spins. This does not

seem to present any practical problem, because the magnetic field of, e.g., a Helmholtz coil or a solenoid enclosing the ³He measurement volume, drops off as distance⁻³ along the off-axis direction. For the cm-scale ³He cell and a similar size coil, this provides ample space for the magnetometer to reside in the proper magnetic

C. Measurement of the ³He dipolar magnetic field with an rf magnetometer

With a $\pi/2$ -pulse, the ³He spins can be tipped to a direction orthogonal to the holding magnetic field and will precess about it with frequency $\omega = \gamma B_h$, where $\gamma = 2\pi \times 3.24 \,\text{kHz/G}$. Let **z** be the unit vector along the common direction of the polarizing magnetic field, the holding magnetic field, and the initial magnetization of 3 He. After the $\pi/2$ -pulse, 3 He spins will precess in the x-y plane (see Fig. 1), and their total magnetic moment will be $\mathbf{m} = M(\mathbf{x}\cos\omega t + \mathbf{y}\sin\omega t)e^{-t/T_2}$, where $M = \mu N_{\text{He}}P_{\text{He}}$, with N_{He} and P_{He} given by Eqs. (1) and (2), respectively, and T_2 being the transverse spin-relaxation time.

Assuming a spherical measurement cell, the dipolar magnetic field produced by a magnetized ³He gas at a distance R away from the measurement cell's center along the x-axis will be $B_{\rm He}e^{-t/T_2}\cos\omega t$, where the magnitude $B_{\rm He}=\mu_0 M/2\pi R^3$, with $\mu_0 = 4\pi \times 10^{-7} \, \mathrm{Tm/A}$ being vacuum's magnetic permeability. This magnetic field is to be sensed by the rf magnetometer. Given that the magnetometer sensor cell is not point-like, we assume an average distance between the center of the sensor cell and the center of the measurement cell $R \approx 3$ cm. Then,

$$B_{\text{He}} = (12 \,\text{aT}) \frac{[B_p/1 \,\text{T}]}{[T_p/100 \,\text{K}]} \left[\frac{\mathcal{C}}{38\%} \right] \left[\frac{\tilde{m}}{1 \,\text{ppb}} \right] \left[\frac{m_s}{200 \,\text{g}} \right].$$
 (3)

The ³He partial pressure in the measurement cell is about 0.2 Torr. We assume that the rest of the gases being released from heating the regolith sample have a mass abundance similar to ³He; thus, we consider a total pressure of 1 Torr in the measurement cell. At such low pressures and for a realistic homogeneity of the holding magnetic field, with gradient at the level of $|\nabla B_h| \approx 1 \,\mathrm{mG/cm}$, the transverse spin-relaxation time of ³He is of the order of 1 h. ⁸⁰ Indeed, at such temperature (100 K) and pressure (1 Torr), the ³He self-diffusion coefficient is⁷⁹ $D \approx 300 \,\mathrm{cm}^2/\mathrm{s}$. The transverse spin-relaxation time is⁸¹ $T_2 \approx 175 D/16 \gamma^2 r^4 |\nabla B_h|^2$, where $r \approx 0.6 \,\mathrm{cm}$ is the measurement cell radius for a measurement cell volume $V_{\rm He}=1\,{\rm cm}^3$. Thus, $T_2\approx 2400\,{\rm s}$.

Currently, rf magnetometers have sensitivities around $\delta B = 1.0 \, \text{fT} / \sqrt{\text{Hz}}.^{61,63}$ Thus, for a measurement time $\tau = 300 \, \text{s}$, one can detect a magnetic field B_{He} at the level of $\delta B/\sqrt{\tau} = 0.06 \, \text{fT}$, which translates to measuring the abundance of regolith-implanted ³He with sensitivity 5 ppb. In other words, the sensitivity, $\delta \tilde{m}$, of the proposed measurement can be expressed as a function of all parameters as

$$\frac{\delta \tilde{m}}{5 \text{ ppb}} = \frac{\left[\delta B / 1.0 \frac{\text{fT}}{\sqrt{\text{Hz}}}\right]}{(12 \text{ aT}) \sqrt{\left[\tau / 300 \text{ s}\right]}} \frac{\left[T_p / 100 \text{ K}\right]}{\left[B_p / 1 \text{ T}\right]} \frac{1}{\left[m_s / 200 \text{ g}\right] \left[\mathcal{C} / 38\%\right]}.$$
 (4)

As a consistency check, the parameter dependence of $\delta \tilde{m}$ makes intuitive sense: $\delta \tilde{m}$ is reduced by reducing δB (increasing the magnetometer sensitivity), increasing measurement time τ , increasing the sample mass m_s (more ³He atoms extracted), reducing the temperature or increasing the polarizing magnetic field (larger ³He spin polarization), and increasing the capture efficiency (more ³He atoms).

We note that the numerical values of the relevant parameters entering Eq. (4) are what we think reasonable and indicative for the workings of this measurement. In an actual realization, several technical design limitations might require different choices for those parameter values. The way the final result is expressed in Eq. (4) can readily accommodate other choices of the parameters and easily lead to the corresponding value for $\delta \tilde{m}$ by inspection. Additionally, the sensitivity expressed in (4) is fundamentally defined by the demand that signal-to-noise ratio (SNR) = 1, although higher values of SNR are practically required for detection. This can be accommodated by choosing different parameter values, such as the measurement time, in order to achieve the desired sensitivity.

IV. DISCUSSION

Here, we wish to discuss the deployability of the proposed measurement in the lunar environment. Regarding apparatus volume, we note that there is steady progress toward developing compact atomic magnetometers; 82-85 thus, the magnetic sensor, including the associated electronics, should not contribute significantly to the volume of the apparatus of Fig. 1. The most voluminous component should be the regolith sample of volume somewhat larger than 100 cm³. The measurement cell of diameter $r \approx 0.6$ cm can be enclosed by a solenoid of similar volume; hence, $\frac{8}{5}$ the holding magnetic field does not significantly contribute to the R volume, which can overall be significantly smaller than 1 l.

Regarding mass, solenoid wire, electronics, heating, and gas-handling systems could contribute 1-2 kg.

Regarding power requirements, there are three major loads. The heating of the atomic magnetometer requires on the order of 100 W, and the electronics associated with the sensor around 100 W (both numbers are exaggerated on the high side). More substantial is the power requirement for the heating of the regolith material. To heat $m_s = 200 \,\mathrm{g}$ of lunar regolith to 1000 K, given the specific heat capacity $c \approx 1 \, \text{kJ/kg/K}$, and the temperature change from 100 K of the lunar night to 1000 K, i.e., $\Delta T = 900$ K, the thermal energy required is $Q = mc\Delta T \approx 180 \text{ kJ}$. Over 300 s, this translates to 600 W. We neglect the power needed to cool the parts of the apparatus required to be at low temperature since the ambient temperature during the lunar night is as low. In total, given an available power of the order of 1 kW, a single prospecting measurement, including the heating phase and the magnetometry phase, takes about 10 min and consumes energy of the order of 100 Wh. The duration could be reduced if more power is available. Equivalently, one could increase the measurement time, thus reducing further the required sample volume according to Eq. (4). This point reiterates our previous comment that the specific parameter values entering Eq. (4) will eventually relate to other mission design parameters, e.g., the available power.

TABLE I. Performance metrics for prospecting for lunar ³He with a radio-frequency atomic magnetometer.

Prospecting specs of an rf magnetometer		
³ He sensitivity	5 ppb	
Regolith sample mass	200 g	
Magnetometric measurement time	5 min	
Total prospecting time	10 min	
Equipment mass	<2 kg	
Equipment volume	~1 l	
Power	~1 kW	
Energy	~100 Wh	
Cost	<\$500 k	

Regarding cost, it is not straightforward to estimate the cost incurred by designing and implementing space-grade hardware realizing the scheme of Fig. 1, but the required equipment altogether should cost significantly less than \$0.5 M if it were to be used in a laboratory.

In Table I, we summarize the aforementioned performance metrics, again, not within the precision of a technical design report for a space mission, but within reasonable estimates based on current laboratory-grade technology.

In summary, the proposed methodology can fit a compact and low-cost design taking advantage of the robust and simple-to-operate modern atomic magnetometers. Thus, such prospecting equipment could be readily mounted on a small lunar rover accompanying the mining infrastructure. One could imagine that such a rover would in short time sample and prospect regolith material from several regions. The actual mining machine would then retrieve regolith material from those regions having the highest abundance of ³He, for example, larger than 20 ppb.

AUTHOR DECLARATIONS

Conflict of Interest

The authors have no conflicts to disclose.

Author Contributions

I. K. Kominis: Conceptualization (lead); Formal analysis (lead); Methodology (lead); Project administration (equal); Writing - original draft (lead); Writing - review & editing (lead). C. Kosmas: Project administration (equal).

DATA AVAILABILITY

Data sharing is not applicable to this article as no new data were created or analyzed in this study.

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METHOD

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